Lecture: 28

TRANSAMINATION, DEAMINATION AND DECARBOXYLATION

- Protein metabolism is a key physiological process in all forms of life.
- > Proteins are converted to amino acids and then catabolised.
- > The complete hydrolysis of a polypeptide requires mixture of peptidases because individual peptidases do not cleave all peptide bonds.
- ➤ Both **exopeptidases** and **endopeptidases** are required for complete conversion of protein to amino acids.

Amino acid metabolism

- > The amino acids not only function as **energy metabolites** but also used as **precursors** of many physiologically important compounds such as **heme**, **bioactive amines**, **small peptides**, **nucleotides and nucleotide coenzymes**.
- ➤ In normal human beings about 90% of the energy requirement is met by oxidation of carbohydrates and fats. The remaining 10% comes from oxidation of the carbon skeleton of amino acids.
- > Since the 20 common protein amino acids are distinctive in terms of their carbon skeletons, amino acids require unique degradative pathway.
- > The degradation of the carbon skeletons of 20 amino acids converges to just seven metabolic intermediates namely.
- i. Pyruvate
- ii. Acetyl CoA
- iii. Acetoacetyl CoA
- iv. α -Ketoglutarate
- v. Succinyl CoA
- vi. Fumarate
- vii. Oxaloacetate
- \triangleright Pyruvate, α -ketoglutarate, succinyl CoA, fumarate and oxaloacetate can serve as **precursors for glucose synthesis** through gluconeogenesis.Amino acids giving rise to these intermediates are termed as **glucogenic**.
- ➤ Those amino acids degraded to yield **acetyl CoA** or acetoacetate are termed **ketogenic** since these compounds are used to synthesize ketone bodies.
- > Some amino acids are **both glucogenic and ketogenic** (For example, phenylalanine, tyrosine, tryptophan and threonine.

Catabolism of amino acids

The **important reaction** commonly employed in the breakdown of an amino acid is always the **removal of its** α -amino group. The product ammonia is excreted after conversion to **urea** or other products and the carbon skeleton is degraded to CO_2 releasing energy. The important reaction involved in the deamination of amino acids is

- i. Transamination
- ii. Oxidative deamination
- iii. Non oxidative deamination

Transamination

- ➤ Most amino acids are deaminated by transamination reaction catalysed by aminotransferases or transaminases.
- \succ The α-amino group present in an amino acid is transferred to an α-keto acid to yield a new amino acid and the α-keto acid of the original amino acid.
- \succ The predominant amino group acceptor is α -**keto glutarate**. Glutamate's amino group is then transferred to oxaloacetate in a second transamination reaction yielding aspartate.

Glutamate + oxaloacetate
$$\xrightarrow{}$$
 α -ketoglutarate + aspartate pyridoxal phosphate

- > Pyridoxal phosphate, the coenzyme of pyridoxine (vitamin B6) plays an important role in these reactions.
- Amino transferase reactions occur in two stages.
- O Pyridoxal phosphate is covalently attached to the amino transferases via a **Schiff's base linkage** formed between the aldehyde group of pyridoxal phosphate and the **epsilon amino group of lysine** residue of the enzyme. Pyridoxal phosphate is converted to pyridoxamine phosphate.
- o In the second stage, the amino group attached to pyridoxamine phosphate is transferred to a different keto acid to yield a new amino acid and releases pyridoxal phosphate

Oxidative deamination

> Transamination does not result in net deamination, since one amino acid is replaced by another amino acid.

- ➤ The function of transamination is to funnel the amino nitrogen into one or a few amino acids.
- For glutamate to play a role in the net conversion of amino groups to ammonia, a mechanism for glutamate deamination is needed so that α -ketoglutarate can be regenerated for further transamination.
- > The generation is accomplished by the oxidative deamination of glutamate by glutamate dehydrogenase.
- ➤ Glutamate is oxidatively deaminated in the mitochondrion by glutamate dehydrogenase. NAD+ or NADP+ functions as the coenzyme.
- ightharpoonup Oxidation is thought to occur with the transfer of a hydride ion from glutamate's α carbon to NAD(P)⁺ to form α -iminoglutarate, which is then hydrolysed to α -ketoglutarate and ammonia.
- ➤ The ammonia produced is then converted to urea in mammals

 Two non-specific amino acid oxidases namely, L-amino acid and D-amino acid

 oxidases catalyse the oxidation of L and D-amino acids utilizing FAD as their coenzymes.

Amino acid + FAD +
$$H_2O$$
 ----- α -Keto acid + NH_3 + $FADH_2$

Non-oxidative deamination

Amino acids such as serine and histidine are deaminated non-oxidatively

The other reactions involved in the catabolism of amino acids are decarboxylation, transulfuration, desulfuration, dehydration etc.

Decarboxylation

- ➤ The decarboxylation process is important since the products of decarboxylation reactions give rise to physiologically active amines.
- > The enzymes, amino acid decarboxylases are pyridoxal phosphatedependent enzymes.
- \blacktriangleright Pyridoxal phosphate forms a Schiff's base with the amino acid so as to stabilise the α -carbanion formed by the cleavage of bond between carboxyl and α -carbon atom.

 \succ The physiologically active amines epinephrine, **nor-epinephrine**, **dopamine**, **serotonin**, γ -amino butyrate and histamine are formed through decarboxylation of the corresponding precursor amino acids.